

Accelerating CO₂ Uptake in Cement Pastes through Nano-TiO₂ Modification

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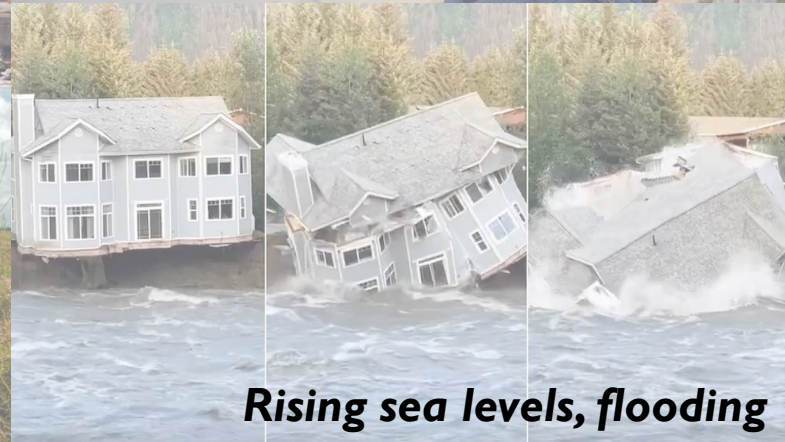
1. Introduction & Motivation
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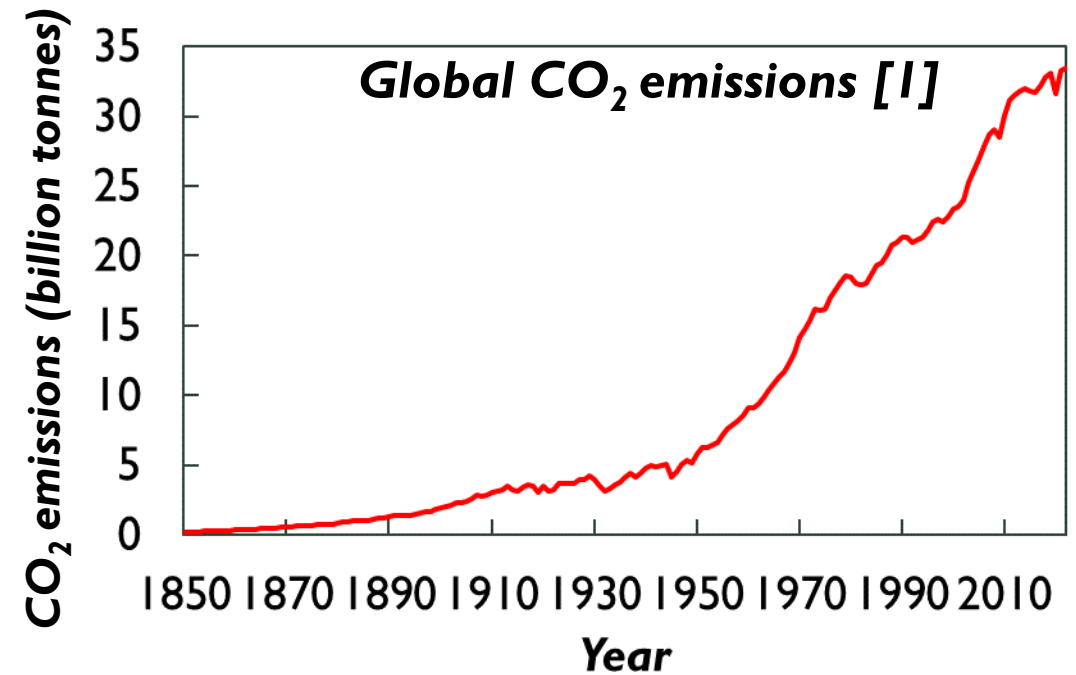
I. Introduction



Climate change



Climate change



Alarming CO₂ rising levels

WHO IS RESPONSIBLE?

Influence of the **construction and buildings sector**





Cement



Water

Aggregates



8% of the total
CO₂ emissions each year [2]

Due to the broad and wide use of
CONCRETE

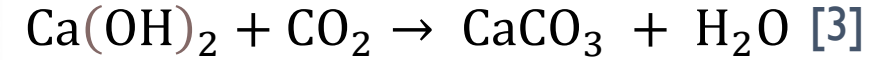
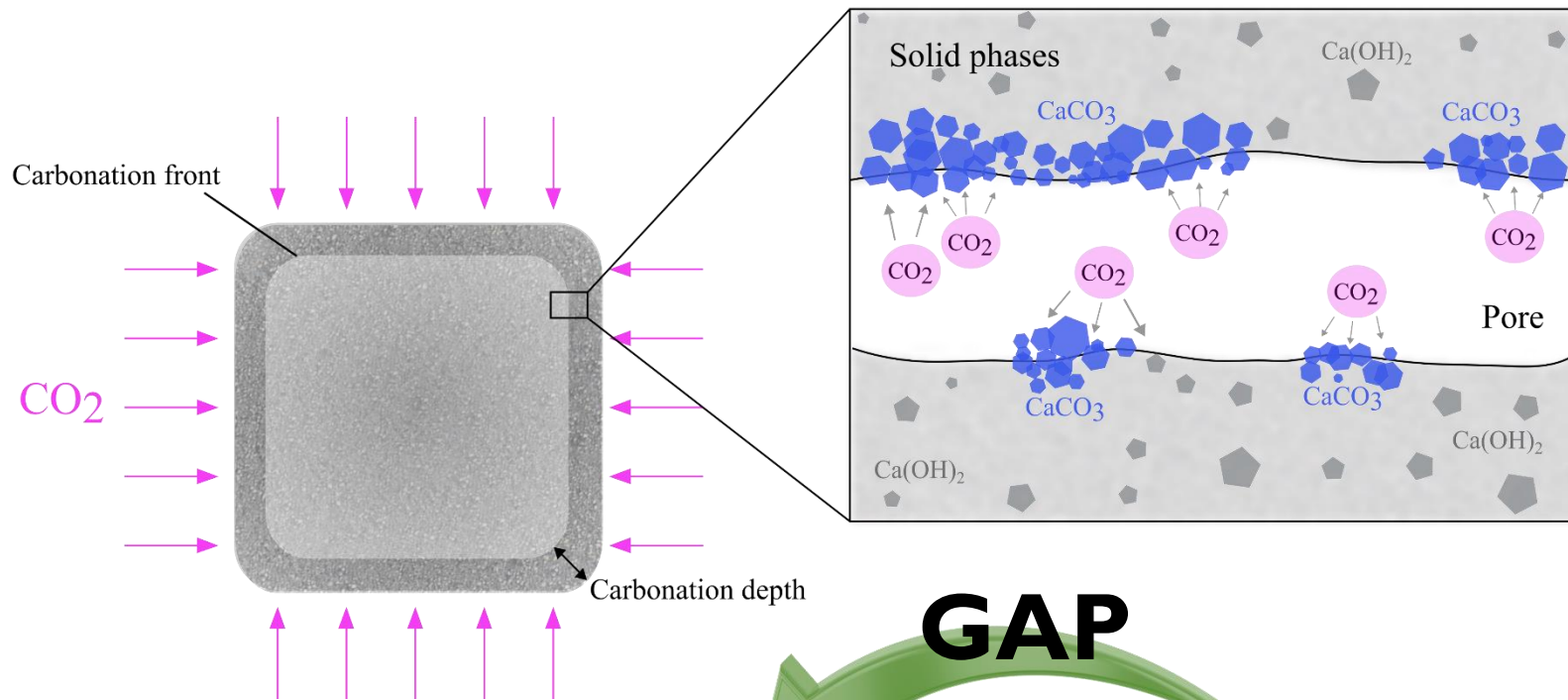


OPPORTUNITY



CONCRETE
PART OF THE SOLUTION

Carbonation: CO_2 sequestration in cement-based materials



Slow and long process



Ways to accelerate it [4]

CO_2 curing

- Increases CO_2 uptake [5]
- Increases compressive strength [6]

Adding nano- TiO_2

- Enhances mechanical properties [8]
- Provides photocatalytic properties [9]
- Increases durability [10]

Our previous study:

- Increases CO_2 uptake [7]



Objective

To understand the fundamental mechanisms that govern the **CO_2 uptake rate variations** produced by **nanomodification** of cement pastes and **CO_2 concentrations**.

GAP

The effect of **nano- TiO_2** and **CO_2 concentration** on the **CO_2 uptake rate is not yet known**

2. Materials & Methods



OPC: Type I



Water

Nano-TiO₂

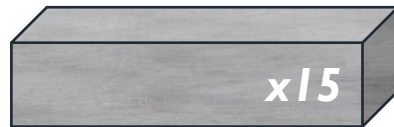
0% and 1%

Cement pastes

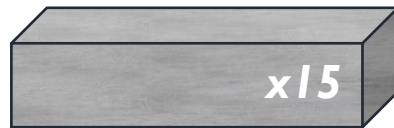
16x4x4 cm³

w/c = 0.55

CP0

0% nano-TiO₂

CPI

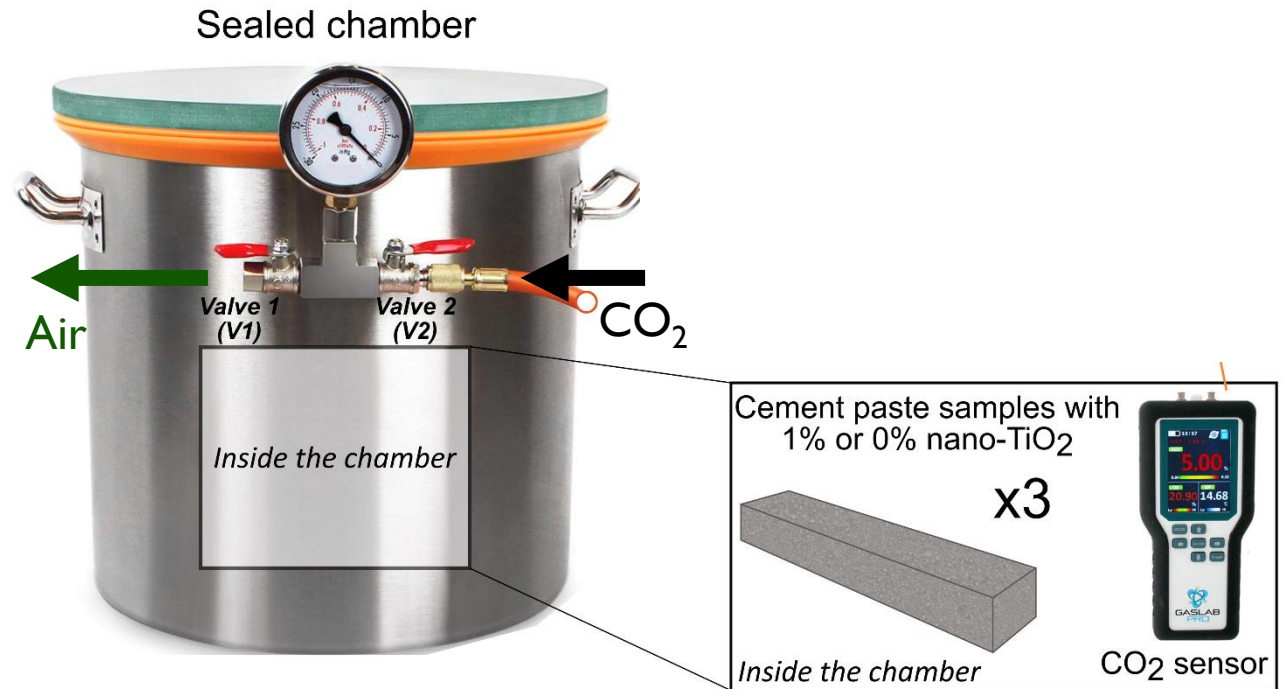
1% nano-TiO₂

Curing procedure

In-mold
(24h)T=23±0.5°C
RH=50±2%In-water
(27d)T=25±1°C
RH=100%Pre-curing
(24h)T=23±0.5°C
RH=50±2%To assess the effect of nano-TiO₂
addition on the CO₂ uptakeAccelerated CO₂ exposure
(8h)

CO₂ exposure

1. 3 samples + a CO₂ sensor are placed inside a sealed chamber



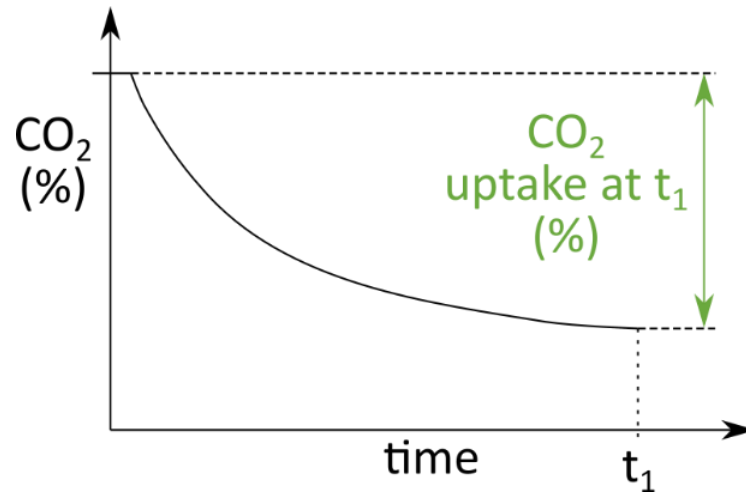
- 25% CO₂
- 50% CO₂
- 100% CO₂

4. 8 consecutive 1-hour cycles were performed (total of 8 hours)
5. Samples were weighted before and after

Estimation of CO_2 uptake and uptake rate

$$\text{CO}_2 \text{ uptake}(\%) = \frac{\rho_{\text{CO}_2} \cdot [V_{\text{chamber}} \cdot (\% \text{CO}_{2,\text{start cycle}} - \% \text{CO}_{2,\text{end cycle}})]}{M_{\text{cement}}} \cdot 100$$

CO₂ sensor monitoring



Before CO₂



After CO₂

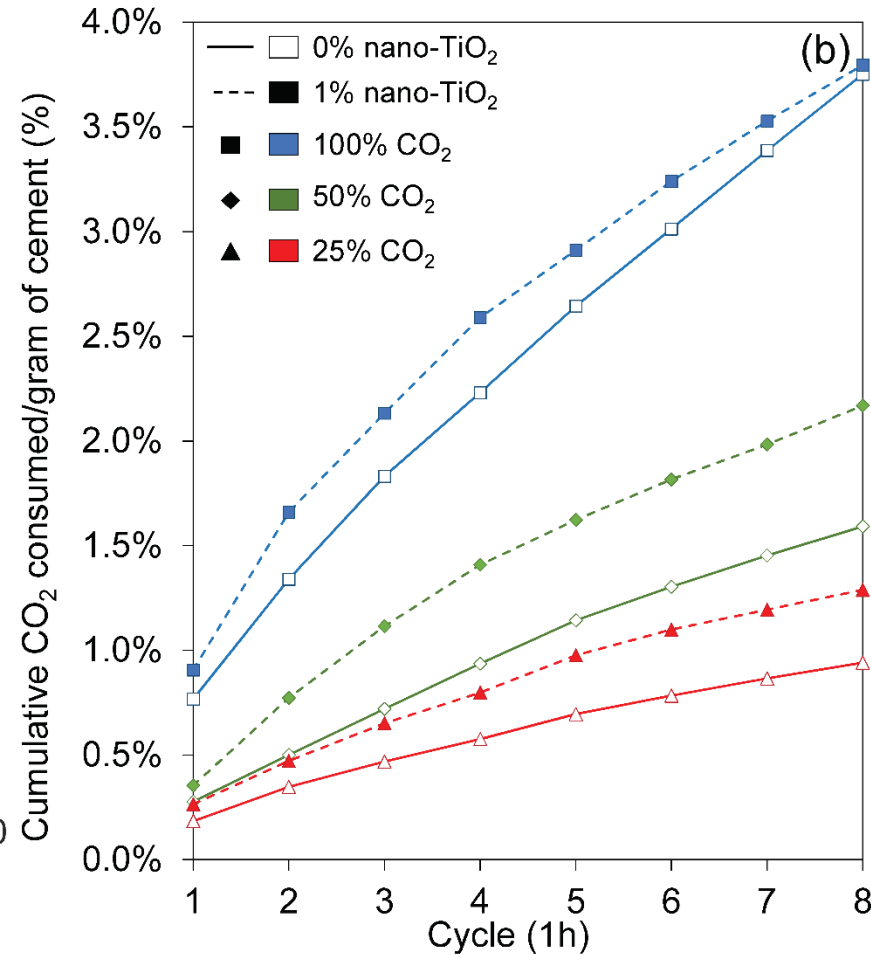
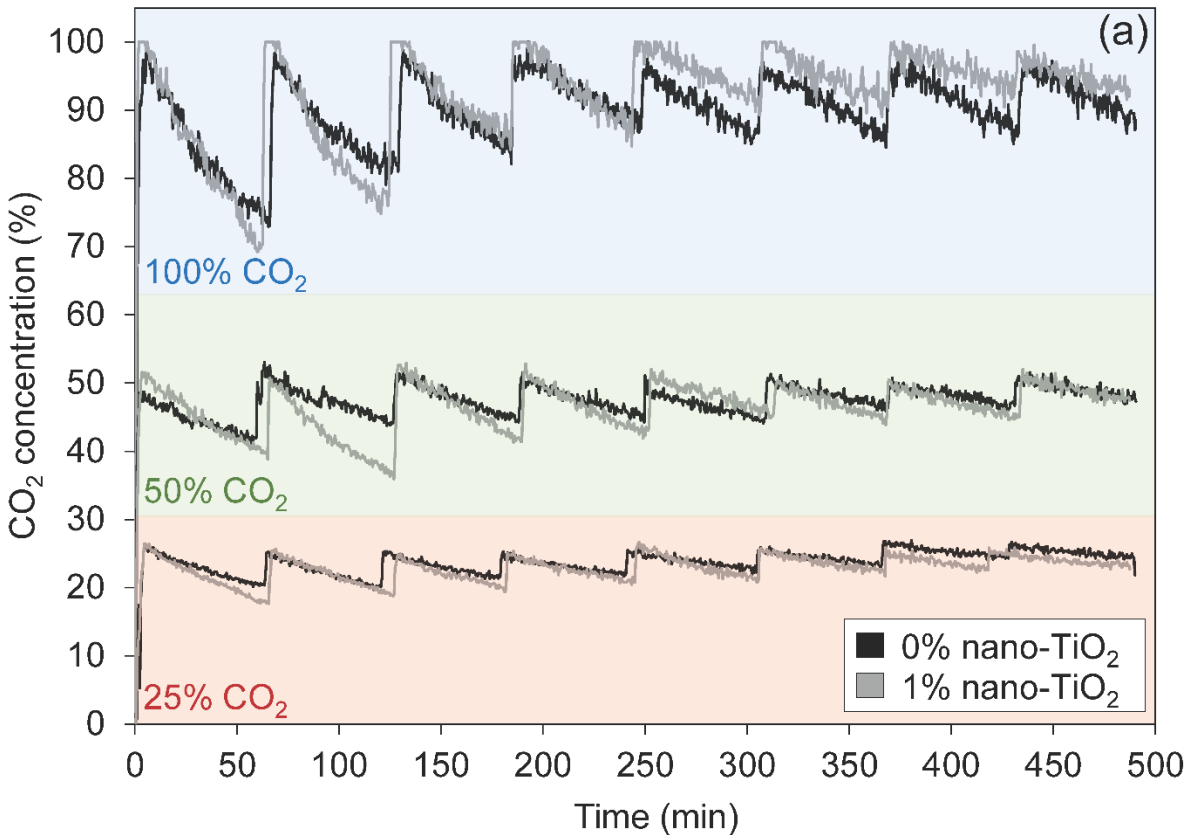


Weight method

$$\text{CO}_2 \text{ uptake}_i(\%) = \frac{M_i + \left(\frac{M_i}{M_T}\right) \cdot M_{\text{water}}}{M_{\text{cement}}} \cdot 100$$

3. Results



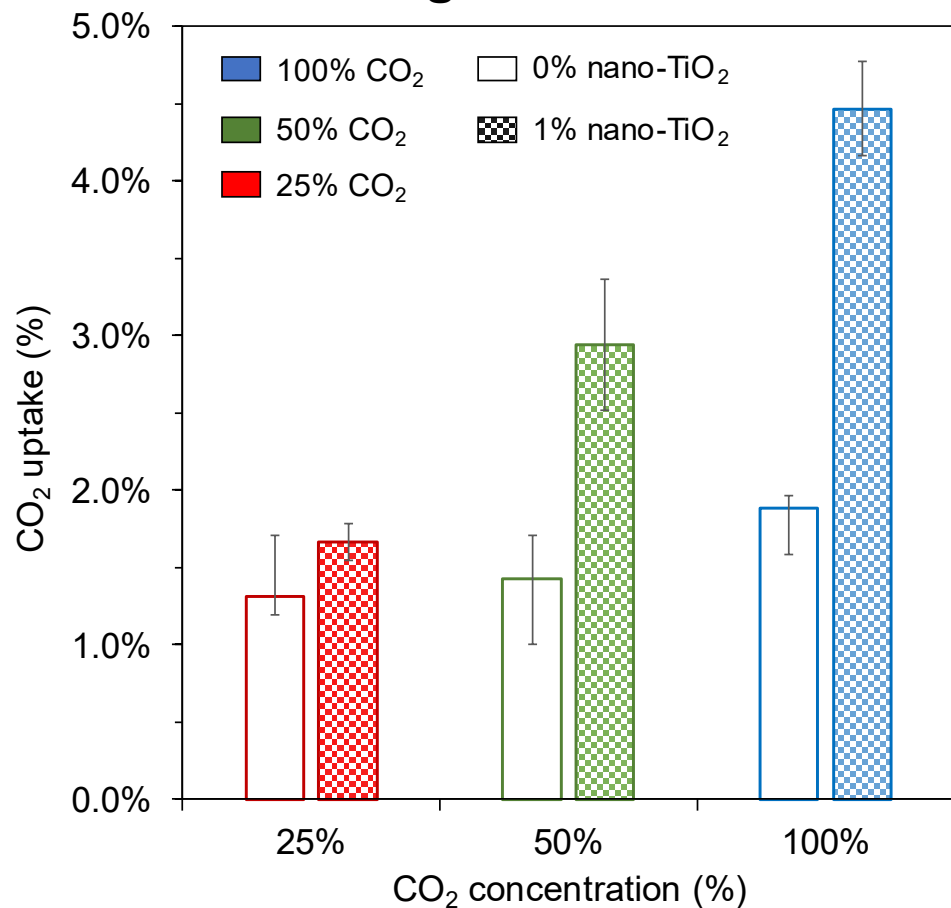
CO₂ sensor monitoring

Nanomodification of the samples enhanced carbonation especially in the first cycles → reduction of surface porosity

- An increase **CO₂ concentration** and adding **nano-TiO₂**, on their own enable **more CO₂ capture at a faster rate**

The contribution of the nano-TiO₂ to the CO₂ uptake rate is higher for lower CO₂ concentrations (25-50% CO₂) than for higher (100% CO₂).

Weight method



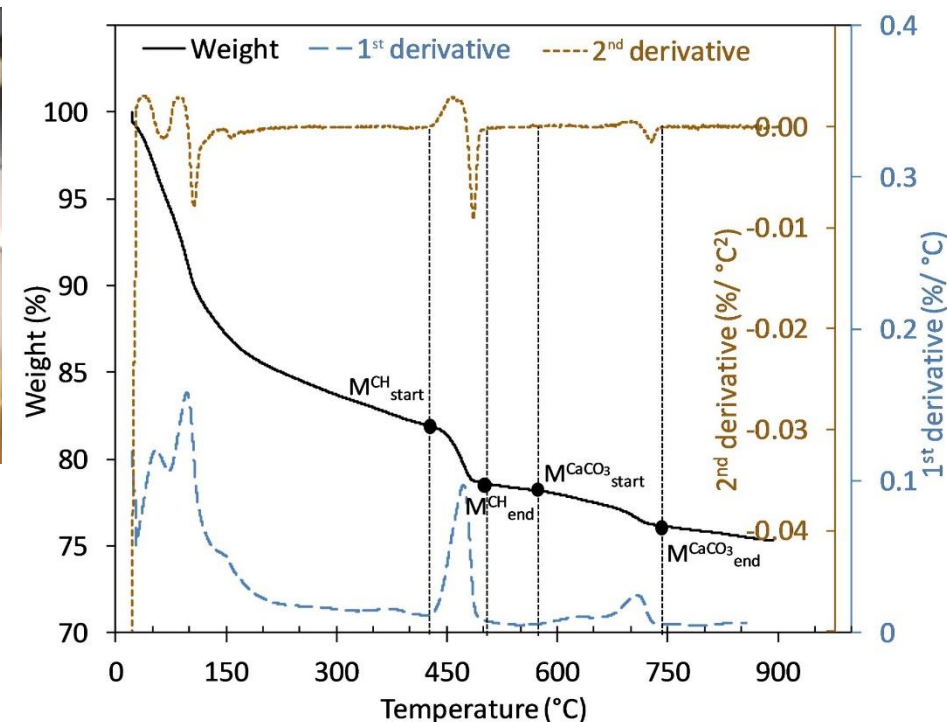
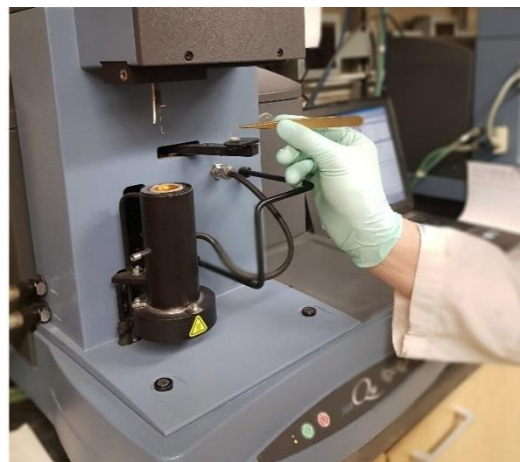
Nano-TiO₂ increased CO₂ uptake in all cases

This method may **underestimate CO₂ uptake** due to water vapor lost during chamber operations

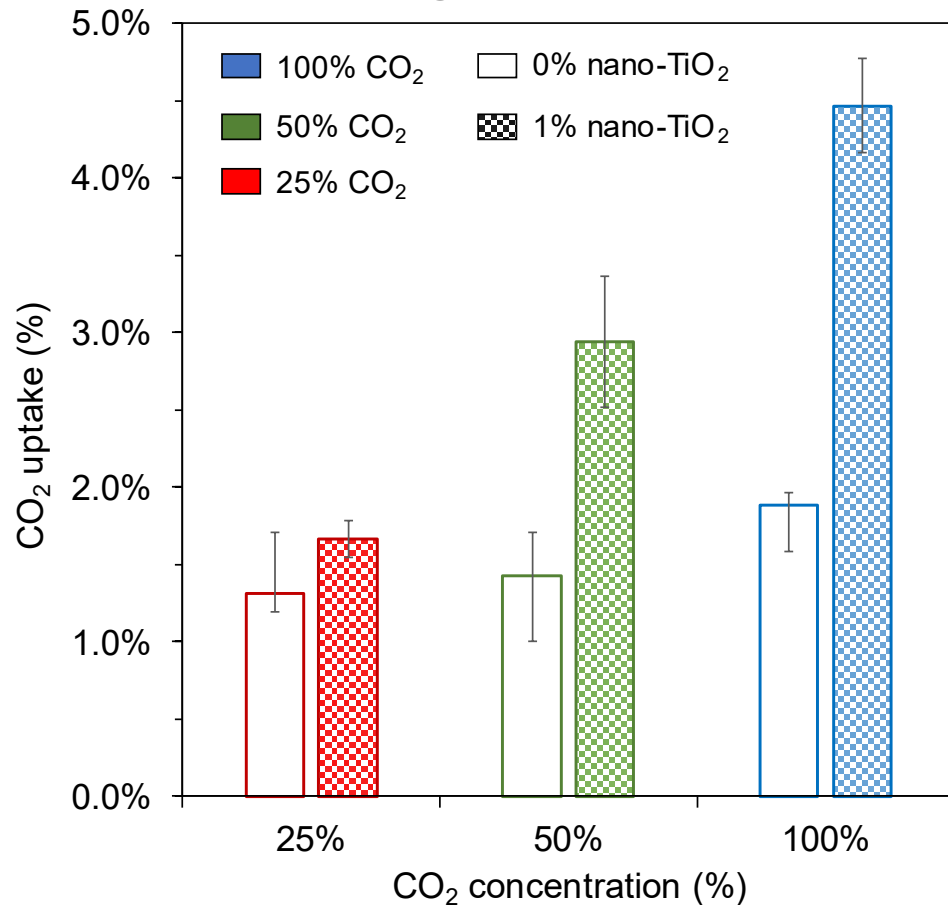
Thermogravimetric Analysis (TGA)

$$CaCO_3(g/100g) = 100 \cdot \frac{100.1}{44.0} \cdot \frac{1}{M_C} \cdot [M_{start}^{CaCO_3} - M_{end}^{CaCO_3}]$$

$$CO_2 \text{ uptake}(g/100g) = [CaCO_3^{C,sample} - CaCO_3^{NC,sample}] \cdot \frac{44.0}{100.1}$$



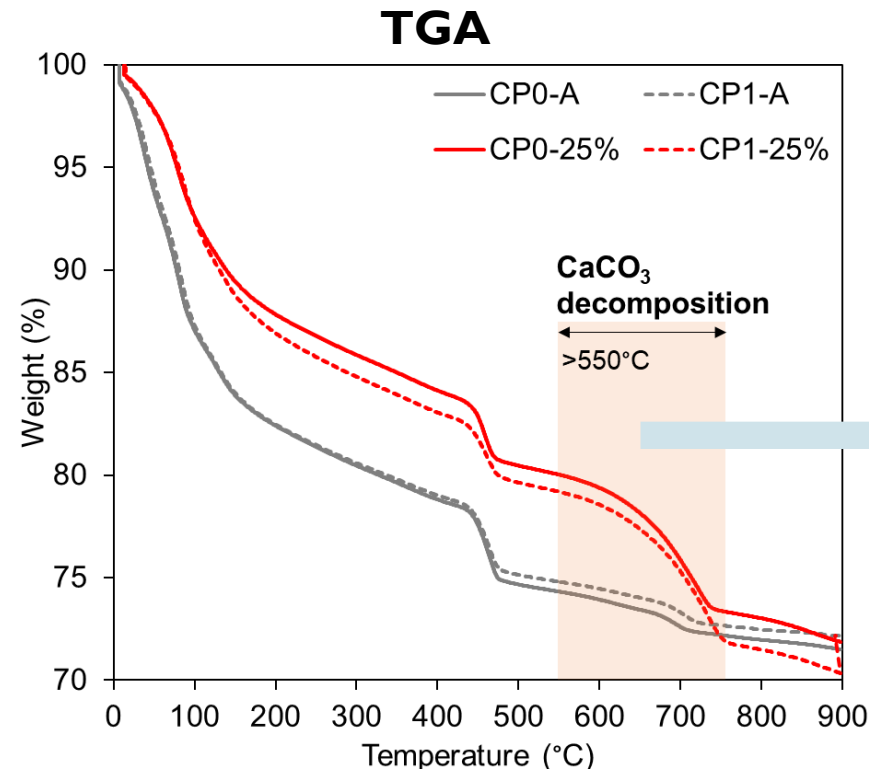
Weight method



Nano-TiO₂ increased CO₂ uptake in all cases

This method may **underestimate CO₂ uptake** due to water vapor lost during chamber operations

Calcium carbonate quantification in samples unexposed and exposed to 25% CO₂



Nanomodification in samples exposed to 25% CO₂ increased **CO₂ uptake** by:

32.7%

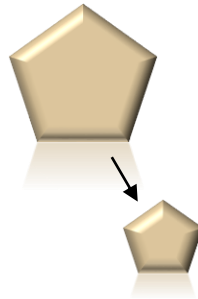
The trend of the effect of **nano-TiO₂ increasing CO₂ uptake** is consistent in all methods

?

Porosity is reduced with using nano-TiO₂ [7,11]

Reduction of
CO₂ uptake

Our results are
counterintuitive



OUR PREVIOUS STUDY:

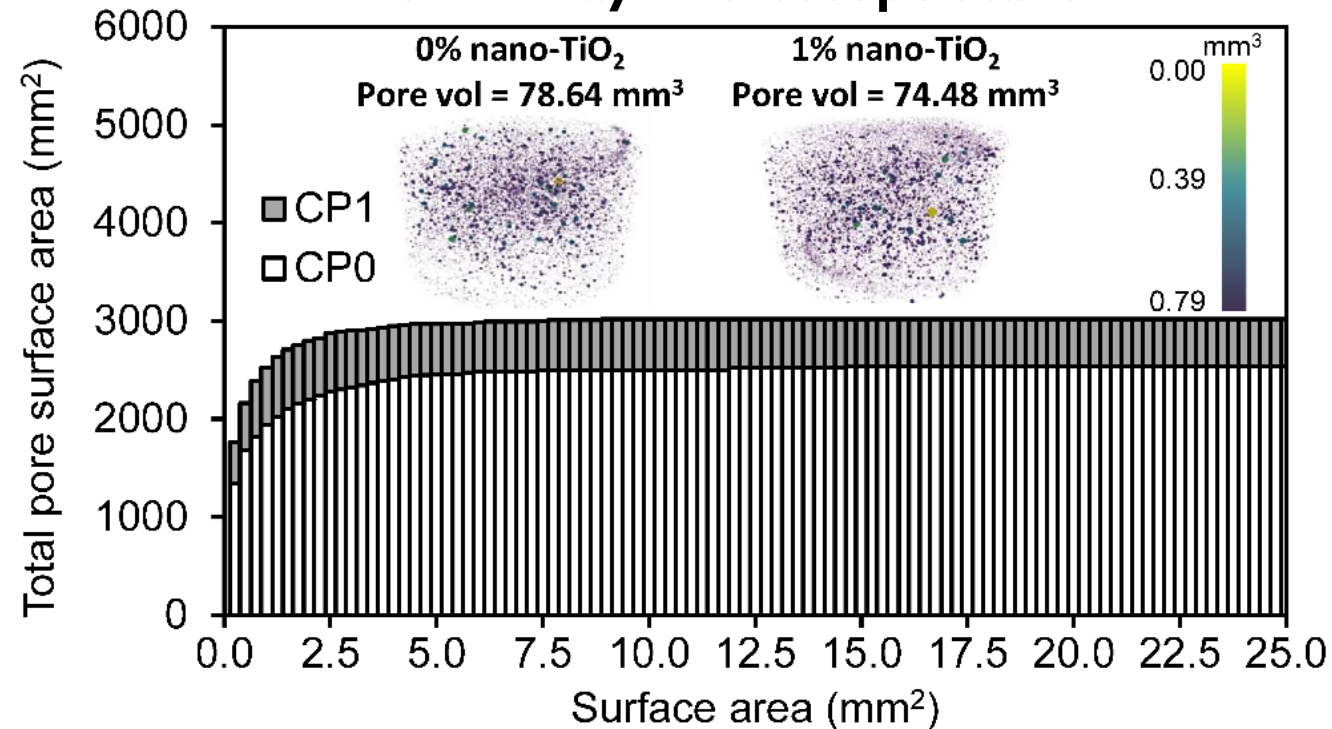
May be due to reduction of the CH
crystal size with nano-TiO₂ [7]

Is there any other mechanism affecting CO₂
uptake?

While porosity is reduced with nano-TiO₂,
**pore surface area of the
nanommodified samples was higher**
than those without nanoparticles

- This might cause the **acceleration and increase of the carbonation** reaction observed
- Higher pore surface → more surface available for CO₂ to react

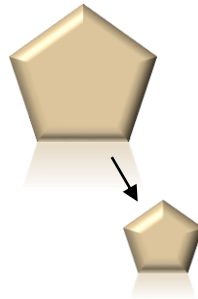
3D X-Ray microscope scans



Porosity is reduced with using nano-TiO₂ [7,11]

Reduction of CO₂ uptake

Our results are **counterintuitive**



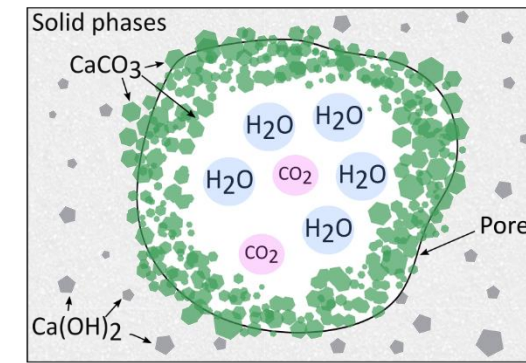
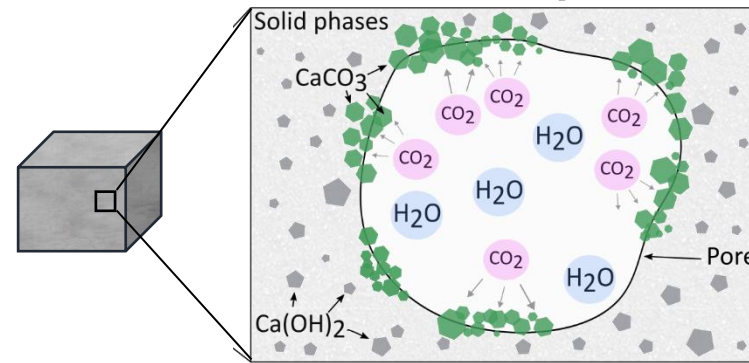
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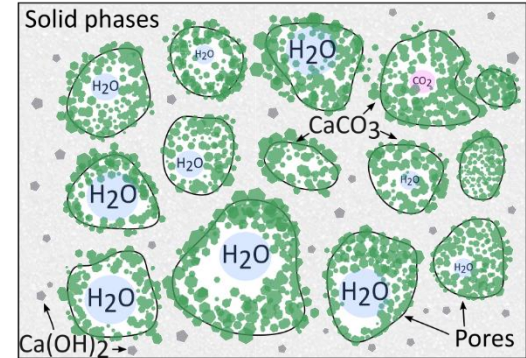
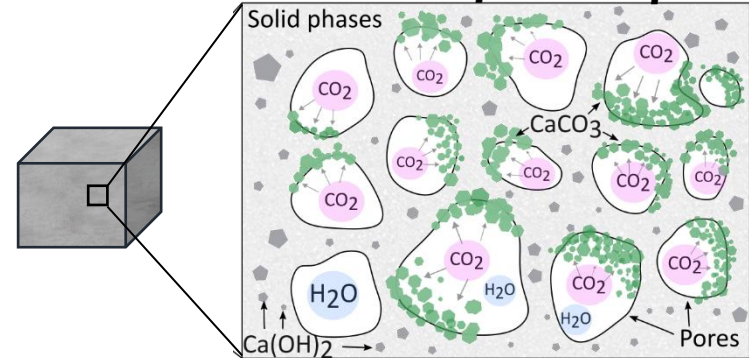
Is there any other mechanism affecting CO₂ uptake?

While porosity is reduced with nano-TiO₂, **pore surface area of the nanomodified samples was higher** than those without nanoparticles

Reference sample



Nanomodified sample



Carbonation

- This might cause the **acceleration and increase of the carbonation** reaction observed
- Higher pore surface → more surface available for CO₂ to react

CO₂ uptake rate estimation model

Carbonation depth: proportional to the square root of time [12]

$$CO_2 \text{ uptake (g)} = A \left(\frac{g}{\sqrt{h}} \right) \cdot \sqrt{t(h)}$$

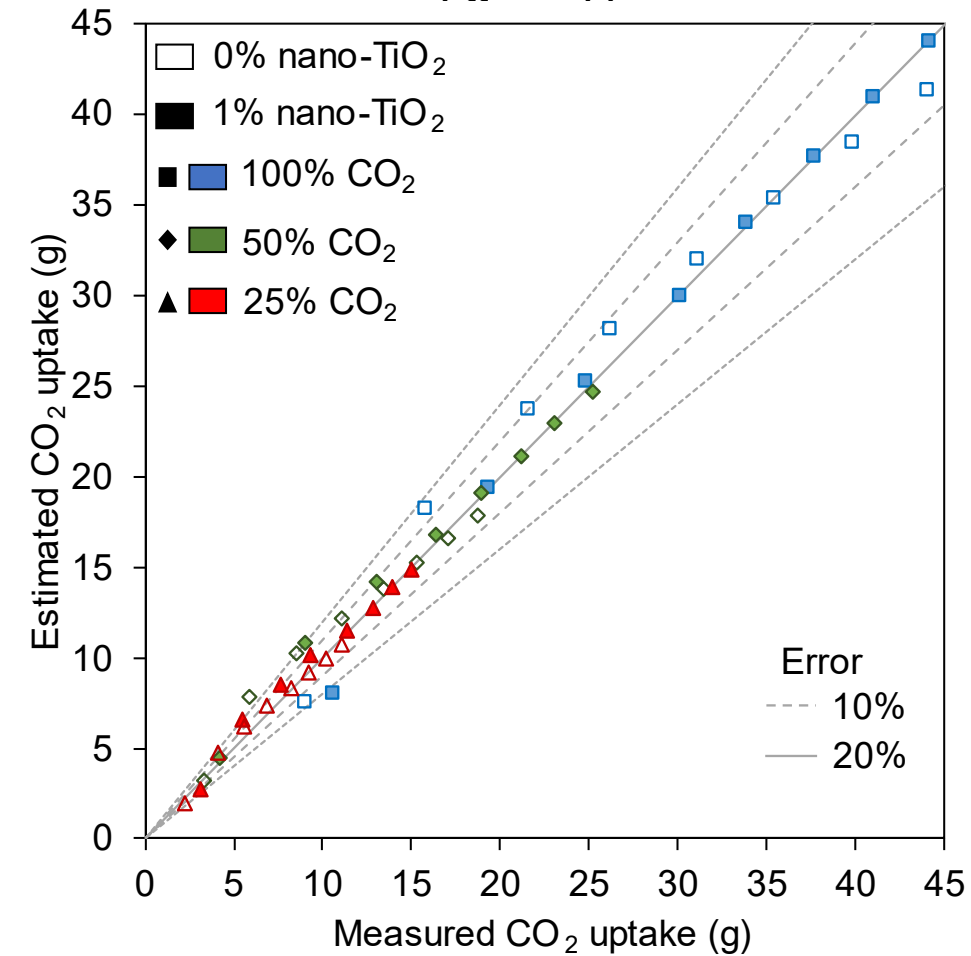
Carbonation rate coefficient

Data from results

One value of **A** (g/h^{0.5}) was estimated for each condition

Least squares method

Condition	Nano-TiO ₂ (%)	CO ₂ (%)
1	1	100
2	0	100
3	1	50
4	0	50
5	1	25
6	0	25



CO₂ uptake rate estimation model

Carbonation depth: proportional to the square root of time [12]

$$CO_2 \text{ uptake (g)} = A \left(\frac{g}{\sqrt{h}} \right) \cdot \sqrt{t(h)}$$

Carbonation rate coefficient

Affected by:

- Nanomodification
- CO₂ concentration

Data from results

One value of **A** (g/h^{0.5}) was estimated for each condition

Least squares method

Condition	Nano-TiO ₂ (%)	CO ₂ (%)	A (g/h ^{0.5})	
1	1	100	8.05	7%
2	0	100	7.55	
3	1	50	4.51	38%
4	0	50	3.27	
5	1	25	2.72	39%
6	0	25	1.96	

- Using 1% nano-TiO₂ increases the carbonation rate coefficient
- The increase is higher for lower CO₂ concentrations

5. Conclusions



Main take-aways from the study

- **CO₂ uptake rate is enhanced** with the use of **nano-TiO₂** for all studied CO₂ concentrations.
- The **effectiveness of nano-TiO₂** addition in terms of CO₂ uptake rate acceleration is **higher with lower CO₂ concentrations** than with 100% CO₂.
- Even though porosity is reduced with nano-TiO₂, **pore surface area of the nanomodified samples was higher** than those without nanoparticles, which might be one of the **responsible mechanisms for the acceleration** of the CO₂ uptake.



Lopez-Arias, M., Moro, C., Francioso, V., Elgaali, H. H., & Velay-Lizancos, M. (2023). **Effect of nanomodification of cement pastes on the CO₂ uptake rate.** *Construction and Building Materials*, 404, 133165.



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Thank you for your attention!

ACI Concrete Convention. March 26th, 2024

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